THE USE OF ARTIFICIAL NEURAL NETWORK TO OPTIMIZE THE pH RESPONSE RANGE OF CHLOROPHENOL RED

F. Wong, M. Ahmad* and L.Y. Heng
School of Chemical Sciences and Food Technology, Faculty Science and Technology, National University of Malaysia, 43600 Bangi, Selangor Darul Ehsan, Malaysia
Received 12 March 2003, Accepted 22 May 2003

ABSTRACT
Artificial Neural Network (ANN) had been used in this study to extend the response range of the pH indicator. The input from absorbance values of the absorbance spectra of chlorophenol red at different pH was used to train the ANN. During the training process, the coefficient values of the ANN will be adjusted to obtain the desire output. In this research, back propagation algorithm had been used for optimizing the response range of the pH indicator chlorophenol red in solution. The result indicates that the use of ANN enable the pH response range to be extended from 4.8-6.8 to 1.0-10.0.

Keywords : Artificial neural network, back-propagation algorithm, pH indicator, chlorophenol red.

1. INTRODUCTION
Chlorophenol red is a pH indicator, which gradually changes its colour over a range called “visual transition interval” 1. Most of the pH indicators determine the pH changes indirectly in limited linear dynamic range, often 2-4 pH units only. Chlorophenol red indicator changes its colour from yellow to purple in a narrow interval of 4.0-7.0 pH units. To solve this problem especially in optical fiber optic pH sensor research, a lot of approach has been used such as using multiple pH indicators2,3, the use of indicators with multiple steps of acid dissociation and ANN technique4-6.

The original works on ANN were published more than 50 years ago by McCulloch and Pitts7,8 and Webb9. Lately, ANN has been shown to provide a superior alternative mechanism for modeling non-linear systems10-14. Application of ANNs is similar to conventional non-linear modeling techniques that require a model-structure design and parameter-estimating cycle to calibrate the model but ANNs use more general modeling approach, i.e. model structure need not to be defined explicitly12.

*Corresponding author-email: andong@pkrisc.cc.ukm.my
In this paper, a multiplayer feed-forward ANN will be applied to a series of absorbance spectra generated from the pH-sensitive indicator solution, chlorophenol red (CPR). The main objective was to extend the pH response range of the CPR pH indicator while at the same time maintaining the prediction error at the acceptable level.

2. EXPERIMENTAL

Apparatus
All the absorption spectra were measured by using Varian Cary 100 UV-VIS spectrophotometer. The pH of the buffer solution was measured with an Ecomet pH/mV/Temp P25 meter.

Reagents
All the reagents were used as received without any further purification. Tris-HCl (0.05 M) buffer solution was prepared by following standard procedure. An appropriate amount of tris (hydroxymethyl) aminomethane (Fluka Chemika) salt was dissolved in deionized water (Water Deionized E-Pure Barnstead). Then, the tris-HCl buffers were prepared in different pH that range from pH 1-14 by adding hydrochloric acid or sodium hydroxide to the buffer solution to decrease or increase the pH, respectively. On the other hand, the pH indicator, CPR solution (2.06 x 10^{-5} M) was prepared by dissolving 4.01mg CPR salt (Riedel-De Haen) in deionized water and the solution was made to 250 mL volumetric flask by using deionized water. Total volumes of 6.0 mL tris-HCl buffer solution and 3.0 mL chlorophenol red solution was added into volumetric flask (10 mL) and deionized water was added into the flask until it reaches the 10 mL level. Then, the absorbance spectrum of each solution was recorded by using UV-VIS spectrophotometer at wavelength range from 350 to 610 nm. The same procedure was repeated for other buffer solution with pH range from pH 1-14. Absorbance value at the wavelength of 375, 433, 486, 530, 575, 595 and 605 nm of each spectrum was selected for the ANN training purpose.

Artificial neural network and training
The training of ANN was carried out using Mathlab, a three layer back-propagation (BP) ANN program running in Microsoft Windows 98 on a 200 MHz personal computer in conjunction with Microsoft Excel 5.0. The inputs consists of seven neuron which corresponding to the absorption values measured at selected wavelength of 375, 433, 486, 530, 575, 595 and 605 nm. Meanwhile, the output layer involves a single neuron representing the pH.

A set of 7 pH (1, 3, 5, 7, 8, 9, 10) of input was introduced to the network and optimization of the network was done by changing the parameter of the network such as the learning rate, amount of hidden layer and number of epochs. After the training, the outputs (pH value) were compared with the actual valued. The purpose of the optimization is to get the closest value of pH between the predicted and the actual value.

The predictions of calibration data was done by introducing a set (pH 4.0, 5.5, 6.0, 6.5 and 8.5) of calibration data to the optimized network and the outputs for the network were evaluated by comparing the predicted value with the actual pH value measured by using the conventional pH
glass electrode. If the outputs value and the actual value were close enough, then the training will be saved. Otherwise, the whole optimization process will be repeated by changing the parameter of the network.

3. RESULT AND DISCUSSION

Determination of training input

Figure 1 shows the three-dimensional and two-dimentional spectra of the chlorophenol red pH indicator measured at the pH range of 1-10. As shown in Figure 1A, the maximum absorption peak changes from 435 nm to 575 nm when pH increased. Figure 1B shows this changes more clearly where when pH increased, the absorption value at 435 nm decreased whereas the absorption value at 575 nm increased. Also, an isobestic point was observed in this spectrum at the wavelength of 480 nm. Figure 2 shows the plot of the indicator absorbance values versus pH values at two different wavelengths of 575 nm and 433 nm. As shown, the absorbance values at 433 nm decreased with increased pH. On the other hand, at 575 nm absorbance value increased with increasing of pH value. All the absorbance value from pH 1-10 was use for ANN training purpose. The training of pH between pH 11-14 was not done because when the absorbance data of pH 11-14 was introduced to the network, the ANN was unable to be optimized well and the error was appear to be very high.

Figure 1 : Three-dimensional (A) and two-dimensional (B) spectra of the chlorophenol red indicator at various pH values.
In many ANN applications, which have been previously reported (4-6), learning rate parameter was not optimized when optimizing the ANN. In general, a high value for learning rate accelerates training, but it can easily lead to oscillations around the minimum. However, setting of learning rate to some low value causes an increase in the number of training epochs. Learning rate near zero results in slow learning and in contrast, a learning rate near 1.0 results in fast learning. Learning is a procedure for modifying the weights and biases of a network. As shown in Figure 3, the sum-squared error (SSE) increased when the learning rate become faster. However, when comparing the output of the network with the actual value of pH for both first and second sets of output, it shows that learning rate of 0.001 produced better prediction when compared to the output. During optimization of the learning rate, the number of epochs and the hidden layer were fixed at 10,000 and 25, respectively.

To optimize the number of neurons in hidden layer, the learning rate was maintain at 0.001 whereas the number of epochs was maintained at 10,000. As shown in Figure 4, the sum-squared error decreased when the number of neuron in hidden layer increased. When the
numbers of neurons in the hidden layer are in the range of 15 to 30 neurons, the output produced better prediction values when compared to the actual pH value. For predictions training data fitting, it was generally observed that the smaller the SSE, the closer the output value to the actual value. However, when using the same SSE value for prediction of calibration data, the predicted value was not close to the actual value.

Figure 4: Sum-squared error versus number of neurons in the hidden layer for ANN with learning rate of 0.001 and 10,000 epochs.

Figure 5 shows the plot of the SSE values at different number of training cycle or epochs. As shown, the SSE decreased when the number of epochs increased. This observation agreed well with the results on optimization of hidden layer, which has been previously reported by Musa et al. [5]. In this study, it was found that if the number of epochs is higher than 10,000 epochs; the less accurate result was produced when this network was used for prediction of calibration data. Therefore, it was concluded in this study that the best prediction was obtained when learning rate of 0.001, 25 neurons of hidden layer and 10,000 epochs were used.

Figure 5: Sum-squared error versus number of epochs for ANN with learning rate of 0.001 and 25 neurons in the hidden layers.
Table 1 and Table 2 show the prediction value for training data fitting and calibration data, respectively. From Table 1, all of the predicted values are close to the actual value with average percentage error of 1.4%. On the other hand, Table 2 also shows that the predicted values for calibration data are close to the actual values and only pH 5.5 shows slightly higher percentage error of 4.3%. The average percentage error is only 1.9%. As shown in Figure 6, the predicted

Table 1: Predictions of training data fitting of the network at learning rate of 0.001, 25 neurons in the hidden layers and 10,000 epochs.

<table>
<thead>
<tr>
<th>Actual pH value</th>
<th>1</th>
<th>3</th>
<th>5</th>
<th>7</th>
<th>8</th>
<th>9</th>
<th>10</th>
</tr>
</thead>
<tbody>
<tr>
<td>Predicted pH value</td>
<td>1.015</td>
<td>3.018</td>
<td>5.042</td>
<td>7.162</td>
<td>8.208</td>
<td>9.075</td>
<td>10.128</td>
</tr>
<tr>
<td>Percentage error (%)</td>
<td>1.5</td>
<td>0.6</td>
<td>0.9</td>
<td>2.3</td>
<td>2.6</td>
<td>0.8</td>
<td>1.3</td>
</tr>
</tbody>
</table>

Table 2: Predictions of calibration data of the network at learning rate of 0.001, 25 neurons in the hidden layers and 10,000 epochs.

<table>
<thead>
<tr>
<th>Actual pH value</th>
<th>4</th>
<th>5.5</th>
<th>6</th>
<th>6.5</th>
<th>8.5</th>
</tr>
</thead>
<tbody>
<tr>
<td>Predicted pH value</td>
<td>4.0094</td>
<td>5.2654</td>
<td>5.9476</td>
<td>6.6939</td>
<td>8.4068</td>
</tr>
<tr>
<td>Percentage error (%)</td>
<td>0.2</td>
<td>4.3</td>
<td>0.9</td>
<td>3.0</td>
<td>1.1</td>
</tr>
</tbody>
</table>

Figure 6: Validation plot of test data prediction for the network with learning rate of 0.001, 25 neurons in the hidden layers and 10,000 epochs.
values of optimized ANN are close to the actual data (measured by pH meter) and it was found that ANN had successfully extended the pH response range of CPR from limited pH range of 4.8-6.8 to determine pH range of 1-10.

4. CONCLUSION

The use of ANN for extending the useful response range of pH indicator chlorophenol red has been demonstrated in this study. The useful response range of the indicator was found to increase from 4.8-6.8 to 1.0-10.0 by using the optimized ANN with learning rate of 0.001, 25 hidden neurons and 10,000 epochs.

ACKNOWLEDGEMENTS

The authors would like to acknowledge the Ministry of Science, Technology and Environment of Malaysia for providing the research grant coded IRPA 09-03-03-006.

REFERENCES